



Research article

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## Thermal history of diamond from Arkhangelskaya and Karpinsky-I kimberlite pipes

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**Abstract.** This work studies and compares the main morphological, structural, and mineralogical features of 350 diamond crystals from the Karpinsky-I and 300 crystals of the Arkhangelskaya kimberlite pipes. The share of crystals of octahedral habit together with individual crystals of transitional forms with sheaf-like and splintery striation is higher in the Arkhangelskaya pipe and makes 15 %. The share of cuboids and tetrahedrohedral is higher in the Karpinsky-I pipe and stands at 14 %. The share of dodecahedrohedral in the Arkhangelskaya and Karpinsky-I pipes are 60 % and 50 %, respectively. The indicator role of the nitrogen-vacancy *N3* center active in absorption and luminescence is shown. Crystals with the *N3* absorption system have predominantly octahedral habit or dissolution forms derived from the octahedra. Their thermal history is the most complex. Absorption bands of the lowest-temperature hydrogen-containing defects (3050, 3144, 3154, 3188, 3310 cm<sup>-1</sup>, 1388, 1407, 1432, 1456, 1465, 1503, 1551, 1563 cm<sup>-1</sup>), are typical for crystals without *N3* system, where in the absorption spectra nitrogen is in the form of low-temperature *A* and *C* defects. The above mentioned bands are registered in the spectra of 16 % and 42 % of crystals from the Arkhangelskaya and Karpinsky-I pipes, respectively. The diamond of the studied deposits is unique in the minimum temperature (duration) of natural annealing. Based on a set of features, three populations of crystals were distinguished, differing in growth conditions, post-growth, and thermal histories. The established regularities prove the multi-stage formation of diamond deposits in the north of the East European Platform and significant differences from the diamonds of the Western Cisurals. The results suggest the possibility of the existence of primary deposits dominated by diamonds from one of the identified populations.

**Keywords:** diamond; photoluminescence; nitrogen; infrared spectroscopy; Arkhangelskaya pipe; Karpinsky-I pipe; diamond deposit

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**Introduction.** Two primary diamond deposits were discovered on the East European Platform: the Lomonosov deposit of the Zolotitskoye kimberlite field and the Grib deposit of the Kepinskoye field (Arkhangelsk region) [1, 2]. The development of these deposits began only in the XXI century, and their diamonds have not yet been sufficiently investigated by modern analytical methods [3-5]. The kimberlite pipes of these deposits are overlain by Carboniferous and cut through the Upper Vendian deposits. The kimberlite bodies are cratered and their denudation rate is insignificant [1, 6]. The areas promising for the discovery of primary deposits are covered by late deposits. Consequently, placer deposits are not predicted in the territory, and the typomorphic features of the diamond are not significant for the direct purposes of prospecting geology and are not studied enough [2, 3]. Some researchers suggest the possibility of finding primary sources of Ural diamonds in the north of the East European Platform. In this regard, the study of diamond typomorphic features is essential for solving applied prospecting problems and in terms of fundamental studies of diamond formation, issues of regional geology, and mineralageny [4]. The study of diamonds is one of the few ways to identify the patterns of deep processes that led to the formation of kimberlite pipes [5, 7, 8].

Diamond has a very stable structure, an elementary chemical composition, and no post-growth homogenization [9-11]. Indicators of changes in growth conditions and post-growth processes remain in the zoning of diamond crystals, including after the imposition of secondary inhomogeneities [12, 13]. Most other deep-seated minerals, except for zircon [14-16], are homogenized during long-term natural annealing and show no growth zoning [13, 17, 18]. Changing environmental conditions are often studied by the zoning of the chemical composition or micro impurities [19-21]. In diamonds, the volume distribution of crystal structure defects is also studied [11, 22-24]. The thermal history of diamonds is determined by the ratio of nitrogen concentrations in the form of the main nitrogen defects  $C$ ,  $A$ ,  $B$  [12, 25]. These defects include one  $C(N)$ , two  $A(N_2)$ , and four  $B(N_4V)$  substituting nitrogen atoms [26]. During natural annealing of diamond, the main nitrogen defects of the crystal structure and complexes of interstitial atoms  $B'$  are transformed [25] in the series  $C(N) \rightarrow A(N_2) \rightarrow B(N_4V) + B'(I_n)$ , where  $V$  is a vacancy,  $I$  is an interstitial carbon atom,  $N$  is a substituting nitrogen atom.

Nitrogen-vacancy defects  $NV$ ,  $H3(N_2V)$ ,  $N3(N_3V)$ , and their ionized analogs are formed during the transformation. At this stage, the formation of the nitrogen–hydrogen defect  $N3VH$  also occurs [27, 28]. Nitrogen-vacancy defects appear as intermediate defects during the transformation of the main ones. Two approaches can be used to determine the thermal history of diamond crystals and the post-growth evolution of the environment. The first one is to study the zoning of individual crystals [11, 29] since the zoning of each crystal reflects some stages of its ontogeny. This approach involves making diamond plates and therefore is not always possible.

The second approach is to study large samples of crystals with the identification of individual diamond populations or trends in their characteristic features [30]. Populations may differ in thermal history, impurity concentration, crystal morphology, and composition of inclusions. The methodological complexity of the second approach lies in the presence of zones and sectors in most crystals. Sequential growth zones differ in the set of impurities and thermal activity, and the superposition of their characteristics significantly complicates the interpretation of the results. The implementation of the second approach implies the need to identify and analyze statistical patterns between various morphological and physical (spectroscopic) characteristics. Thus, three groups of crystals with different morphological and spectroscopic characteristics were previously identified in diamonds from the commercial Lomonosov deposit [4]. Further study of the morphology and physical characteristics of diamonds from these deposits is important both for understanding the patterns of diamond formation in different regions and for solving the problem of searching for buried sources, as well as for comparison with the Timan and the Urals diamonds.

The Arkhangelskaya and Karpinsky-I kimberlite pipes contain many yellow crystals [3, 30]. The yellow color is due to  $C$  or  $N3$  defects, of which  $N3$  are active in luminescence [31]. The  $C$  and  $N3$  defects correspond to different conditions of crystal annealing and, accordingly, are indicators of their thermal history. Identification of these defects makes it possible to distinguish crystals from different stages of diamond formation. The main objective of the study is a comprehensive non-destructive study of diamond crystals from the Arkhangelskaya and Karpinsky-I pipes and the determination of the thermal history of individual diamond populations.

**Methodology.** In 2020 the mineralogical features of 350 crystals of the Arkhangelskaya pipe and 300 crystals of the Karpinsky-I pipe from industrial mining were studied. A comprehensive spectroscopic study of 164 crystals from the Arkhangelskaya pipe and 152 crystals from the Karpinsky-I pipe was carried out. The size of the crystals varies in the range of 3-5 mm. IR absorption spectra were recorded on a VERTEX-70 (Bruker, Germany) Fourier spectrometer with a Hyperion1000 microscope at a resolution of  $2 \text{ cm}^{-1}$ . The total nitrogen concentration ( $N_{\text{tot}}$ ) and the proportion of nitrogen in the form of  $B$  defects ( $N_{\text{BS}}$ ) were calculated using the known proportionality coefficients [25]. The absorption coefficients of the bands  $B'$  ( $\alpha_{B'}$ ),  $3107 \text{ cm}^{-1}$  ( $\alpha_{3107}$ ) were determined. The PL spectra at room temperature were recorded on a Horiba FL-3 spectrometer with Xe excitation by a 450 W lamp, excitation ( $\lambda_{\text{EX}}$ ) at 350 and 450 nm, with spectral slit widths of the excitation monochromator



and detecting monochromator of 10 and 1 nm, respectively. The spectra at the temperature of 77 K, excited by 488 and 787 nm lasers  $\lambda_{\text{EX}}$ , were recorded on an InVia (Renishaw) spectrometer. Absorption spectra in the range of 200-800 nm with a spectral slit width of 1 nm were recorded on a UV-2550 spectrometer (Shimadzu, Japan) at room temperature. The absorption coefficient at a peak of 415 nm was determined relative to the background line ( $a_{N3}$ ) 450-420 nm with linear approximation. The minimum determined value of the absorption coefficient in the zero-phonon  $N3$  line was  $0.01 \text{ cm}^{-1}$ ; the color is affected by the  $N3$  system at an absorption coefficient of more than  $0.2 \text{ cm}^{-1}$ .

The main goal of studying the morphology of crystals was to identify the growth form for further analysis of their thermal history: octahedra, cuboids, and individual crystals of mixed habit were identified. To determine the growth form of crystals with signs of strong dissolution, it is necessary to make diamond plates or study surface cathode luminescence. Crystals whose growth form is not unambiguously detected by the methods used were analyzed as a separate group.

**Discussion.** A distinctive feature of diamonds from the Lomonosov deposit, in comparison with the well-studied diamond from the primary sources of the Siberian Platform [30], is the high content of curved dodecahedroids and tetrahexahedroids [3]. These types can be very close in appearance since they represent the limiting dissolution form of both cuboids and octahedral crystals. Many of these crystals are distinguished by bulk yellow and yellow-green luminescence; according to the classification of Yu.L.Orlov [32], they can be attributed to the second variety. There are 13 % of such crystals in the Arkhangelskaya pipe, and 25 % in the Karpinsky-I pipe. These values refer to the diamond mined in 2020 and are fully consistent with previously published data [2, 3]. For many of the studied crystals assignment to I or II types of J.L.Orlov's classification is debatable (Table 1).

In the entire sample, only one crystal of IV type of Orlov's classification was found, and it was a crystal from the Arkhangelskaya pipe. This is a fragment of an indefinite habit with a well-defined gray, fibrous, thin, but opaque rim. Eight dark gray and black individual crystals from the Arkhangelskaya pipe sample are conditionally assigned to V type, based on the presence of a more translucent central zone, which can be seen in the chip, and a higher concentration of black fine inclusions in the peripheral zone. The luminescence of such crystals is visually very weak. In the IR and visible regions, the light absorption by such crystals exceeds the range of optical density detected by the spectrometers used. Typical crystals in terms of morphology with different dissolution degrees are shown in Fig. 1. The top two rows represent the octahedral-dodecahedroid morphological series with increasing degree of dissolution, and the bottom one – cuboid – tetrahexahedroid – dodecahedroid.

In the studied collection, dodecahedroids (50-60 %) are sharply dominated by crypto-laminar and fine-laminar, smooth-faced Uralic type and with sheaf-like and sheaf-splintery striation. Characteristic of the deposit are crystals with contrasting concentric striation. Interpretation of such striation requires additional studies; it may correspond to contrasting growth zoning or post-growth plastic deformation. There are single flat-faced sharp-edge octahedra without traces of dissolution in Karpinsky-I and Arkhangelskaya pipes. The share of crystals of octahedral habit is less than 10 %. Together with crystals with sheaf-like and splintery striation near the edges of the octahedra, their share reaches 15 %. Surfaces



Fig. 1. Typical in terms of morphology crystals of Arkhangelskaya and Karpinsky-I kimberlite pipes with different degrees of dissolution

of a different area with sheaf-like and splintery striation form a typical habit of crystals of transitional forms. Crystals with parallel striations are not found.

Table 1

**Mineralogical classification of diamonds by Yu.L. Orlov, frequency of occurrence of crystals by type and habit, %**

Color	A.P.Karpinsky-I pipe (300 samples)								Total by color	
	Type I				Types II+III			Type V	Type VII	
	Octahedra	Mixed-habit diamond	Dodecahedroid	Shapeless crystal fragments	Tetrahedroid	Tetrahedroid+cube	Cube			
Colorless	0.0	3.2	24.8	2.7	1.0	1.4	1.4	0.0	0.0	34.4
Yellow	2.7	3.7	11.4	0.9	5.1	5.5	0.7	0.0	0.0	30.2
Gray-yellow	1.8	1.4	2.7	1.8	5.0	2.7	0.0	0.0	0.0	15.5
Brown	0.7	0.7	7.1	2.3	0.3	1.8	0.3	0.0	0.0	13.2
Yellow-brown	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0
Pink	0.0	0.0	0.7	0.0	0.0	0.0	0.0	0.0	0.0	0.7
Gray	0.3	0.7	2.2	0.0	0.0	0.0	0.0	0.3	0.0	3.5
Dark gray	0.0	0.3	0.7	0.0	0.0	0.0	0.0	0.7	0.0	1.7
Pigmentation patches	0.0	0.0	0.7	0.0	0.0	0.0	0.0	0.0	0.0	0.7
<b>Total</b>	5.6	9.9	50.3	7.8	5.6	11.4	6.1	1.0	0.0	100.0
Arkhangelskaya pipe (350 samples)										
Colorless	2.4	2.8	22.0	1.2	0.0	0.4	0.0	0.0	1.2	29.9
Yellow	3.9	1.6	8.7	0.8	5.9	3.5	0.4	0.0	0.4	25.2
Gray-yellow	0.0	0.0	2.0	0.0	1.2	0.0	0.0	0.0	0.4	3.9
Brown	2.0	0.8	4.7	0.8	0.4	0.0	0.0	0.0	0.0	8.7
Yellow-brown	0.0	0.4	2.0	0.0	0.0	0.4	0.0	0.0	0.0	2.8
Pink	0.0	0.0	0.8	0.0	0.0	0.0	0.0	0.0	0.0	0.8
Gray	0.0	1.2	15.7	0.8	0.0	0.0	0.0	2.0	3.9	23.6
Dark gray	0.0	0.0	2.0	0.0	0.0	0.0	0.0	1.2	0.0	3.1
Pigmentation patches	0.0	0.0	1.2	0.0	0.0	0.0	0.4	0.0	0.4	2.0
<b>Total</b>	8.3	6.7	59.1	3.5	7.5	4.3	0.8	3.1	6.3	100.0

The color is dominated by gray crystals with numerous graphite inclusions that affect the perception of color. Among pure colors, crystals with a yellow hue of varying intensity dominate. Among the fantasy colors, yellow prevails; purple and pink crystals, not typical for the primary deposits of the Siberian Platform, are less common. In the pipes of the Lomonosov deposit, the share of brown crystals is less than 10 %.

Crystals of an octahedral habit in equal proportions are colorless, yellow and yellow with a gray-brown hue. Among the mixed-form crystals, colorless and yellow ones are equally common; brown and grayish-yellow ones are found in single cases. Dodecahedroids in the Arkhangelskaya and Karpinsky-I pipes make up the majority (15 %) of colorless crystals with a high degree of transparency, about 5-7 % are yellow and straw-yellow diamonds with different hue intensity, while 5 % are yellow with a gray (smoky) component in coloration. Tetrahedroids have a specific color: yellow with a low degree of transparency, and yellow with a smoky or gray-green component. Their transparency is lowered due to scattered finely dispersed inclusions, which cannot be detected by non-destructive methods [33], however, tetrahedroids and crystals close to them demonstrate different degrees of saturation with inclusions. Tetrahedroids are characterized by the presence of intense



yellow-green PL, usually uniformly distributed in the volume, but there are samples in which the luminescence is visually determined as surface one.

When carrying out a morphological description of crystals for the subsequent analysis of spectroscopy data, the most important parameter is the identification of the dominant growth mechanism. There are two main growth mechanisms: tangential of  $\{111\}$  faces and normal growth of subparallel  $\{100\}$  surfaces. In the first case, flat-faced octahedra are formed, and in the second – cuboids. Sometimes there are crystals of mixed habit, with joint growth of  $\{111\}$  faces and subparallel  $\{100\}$  surfaces. Crystals the central part of which grew by the normal mechanism and outer zones – by the tangential one are often observed. The difference in the growth mechanism is the basis for the diversity of crystal morphology and its spectroscopic characteristics. In growth pyramids  $<100>$  formed during the growth of subparallel  $\{100\}$  surfaces,  $a_{3017}$  is systematically higher and  $N_{tot}$  is lower than in synchronous zones of  $<111>$  pyramids [22, 23]. Pyramids  $<100>$  in crystals of mixed habit often contain submicron inclusions, in which methane and graphite are detected [34].

The basic parameter that determines the spectroscopic characteristics of diamond crystals is the concentration of nitrogen impurity.  $N_{BS}$ ,  $\alpha_{B'}$ , and  $\alpha_{3107}$  parameters depend on this parameter, as well as the thermal history, and the crystal growth mechanism. A feature of the studied samples is the frequent occurrence in the spectra of two absorption systems, the nature of which has not yet been defined [4]. These systems are revealed in the spectra if there is no  $B'$  band in the absorption and there is no  $N3$  absorption. The first system consists of narrow 3050, 3144, 3154, 3188, and  $3310.5\text{ cm}^{-1}$  lines, along with them, in most cases, the  $1363\text{ cm}^{-1}$  band is recorded. The second includes lines 1388, 1407, 1432, 1456, 1465, 1503, 1551,  $1563\text{ cm}^{-1}$  bands. The systems are registered in the spectra of 14 and 2.5 % crystals of the Arkhangelskaya pipe and 14 and 28 % of diamonds of the Karpinsky-I pipe. Previously, these systems were described in single crystals as a rare feature – the lowest temperature form of hydrogen-containing defects [27]. At the same time, the studied set of samples does not contain crystals of Ib type (without  $A$  defects) or Iab, in which  $C$  defects are the dominant form of nitrogen. In the studied yellow crystals, the concentration of  $C$  defects is below the detection limit of IR spectroscopy. The absorption of the  $N3$  defect (Fig. 2, a) is detected in the spectra of 35 % of the crystals of the Karpinsky-I pipe and 47 % of the Arkhangelskaya pipe. The distribution of crystals by  $a_{N3}$  is lognormal (Fig. 2, b). The  $B'$  band is present in 78 % and 62 % of the spectra of crystals from the Arkhangelskaya and Karpinsky-I pipes, respectively. It is detected in the spectra of 44 % of crystals without  $N3$  in the absorption from the Arkhangelskaya pipe and 43 % from the Karpinsky-I pipe.

Among the studied crystals from the Karpinsky-I pipe, there are no nitrogen-free ones; the maximum value of  $N_{BS}$  is 65 % (Fig. 3, a). Most of the crystals are located in the range of 600-1400 ppm  $N_{tot}$ . The crystal arrangement on the Taylor diagram ( $N_{BS}$ - $N_{tot}$  parameters) is very compact: more than 90 % have an  $N_{BS}$  value of less than 30 %. The distribution of  $N_{tot}$  is unimodal, with a maximum in the range of 1050-1200 ppm.

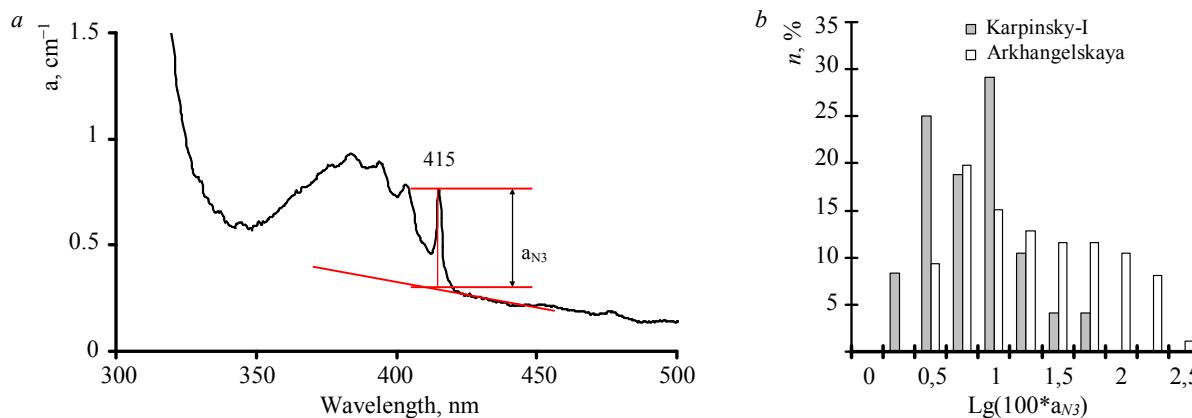


Fig. 2. The absorption spectrum of a crystal with the  $N3$  system (a) and distribution of diamond crystals from the Karpinsky-I and Arkhangelskaya pipes (b) according to the absorption coefficient of the  $N3$  system in the absorption spectrum

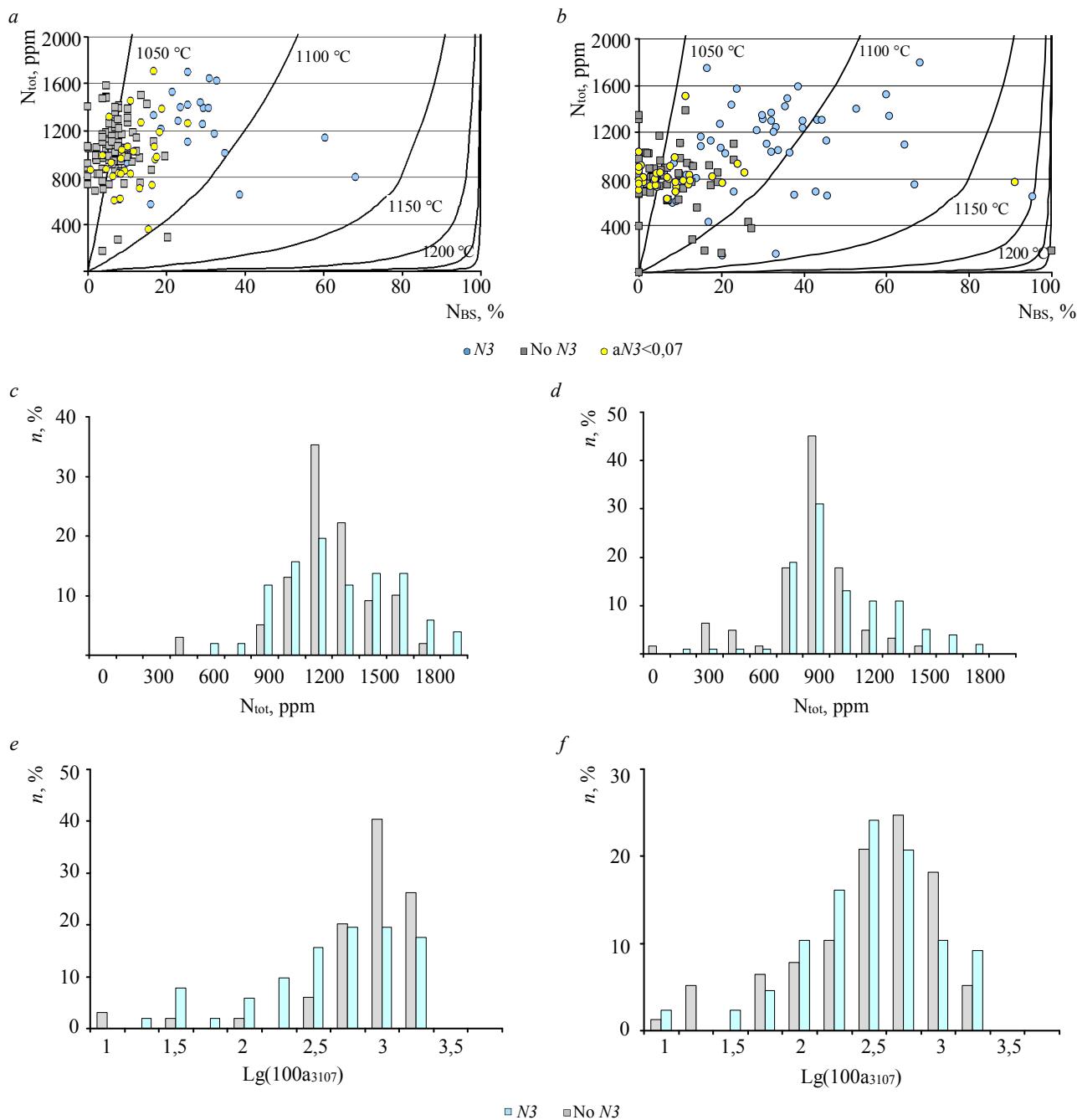


Fig.3. Distribution of crystals of A.P. Karpinsky-I (a, c, e) and Arkhangelskaya (b, d, f) diamond pipes: on Taylor diagram (a, b), model time 3Ga, on total nitrogen concentration (c, d), on the absorption coefficient of  $3107\text{ cm}^{-1}$  band  $\text{N}_3\text{VH}$  defect (e, f) with the division of crystals by the presence of  $\text{N}3$  in the absorption spectrum

Among the crystals of the Arkhangelskaya pipe, one is nitrogen-free and one with an NBS value of 100 % (Fig.3, b). The PL spectrum of this crystal exhibits lines at 536 and 575 nm, which are characteristic of IaB type crystals with signs of plastic deformation. The crystal distribution of the Arkhangelskaya pipe in N<sub>tot</sub> is more compact (Fig.3, c, d), and the crystal distribution field in the Taylor diagram is wider in both N<sub>BS</sub> and N<sub>tot</sub> than for crystals from the Karpinsky-I pipe. The distribution of N<sub>tot</sub> is unimodal, with a maximum in the range of 900-1050 ppm. The crystals of both pipes with the N3 system in the absorption spectra have a higher N<sub>BS</sub> value and a broader N<sub>tot</sub> distribution. The distribution fields on the Taylor diagram of crystals with and without the N3 system overlap, but only single crystals without the N3 system fall above 20 % NBS. On the frequency distribution over the absorption coefficient of the  $3107\text{ cm}^{-1}$  band of the N<sub>3</sub>VH defect, the maximum in crystals



with the  $N3$  absorption system is characterized by a smaller value than in crystals without this defect (Fig.3, *e, f*). This regularity is valid both for the crystals of the Arkhangelskaya and Karpinsky-I pipes. The obtained results show a significant difference between the diamond of these pipes and the diamond-bearing objects of the Urals [35]. Diamond from the alluvial placers of the Western Cisurals is characterized by a distribution over the entire Taylor diagram, there are also low-nitrogen crystals, many crystals with high NBS. The diamond of the Rassolninskoye deposit has a unimodal distribution in  $N_{tot}$  of about 20-40 % NBS, while in the diamond of the studied collections, the main part of the distribution falls in the region of 0-20 %.

All crystals of the octahedral habit from the Karpinsky-I pipe have the  $N3$  system in the absorption spectra (Fig.4, *a*). In the Arkhangelskaya pipe, most of the crystals of the octahedral habit have the  $N3$  system in the absorption spectra, the other three have  $N_{tot}$  less than 300 ppm (Fig.4, *b*). It appears that the concentration of  $N3$  defects in these three crystals is below the detection limit according to the method used because of the low value of  $N_{tot}$ .

The PL spectra of most crystals at room temperature and UV excitation contain the  $N3$  system or a broad amorphous band with a maximum at about 550 nm ( $S3$ ) (Fig.5).

Crystals have blue or green PL. In addition to these bands, the  $S2$  system with a zero-phonon line at 489 nm (poorly detected against the background of the  $S3$  system), a group of lines at 583 + 605 + 627 nm, and a line at 787 nm are recorded. In crystals with the  $N3$  system, it appears in the absorption spectra of 98 % of crystals and dominates in the spectra of 70-80 % of them. In crystals without the  $N3$  system in the absorption, it appears in the 40-55 % of PL spectra (Table 2). The PL spectra of the studied crystals are much more diverse at a temperature of 77 K and an excitation of 488 nm. The main systems observed under these conditions were described earlier [3, 4].

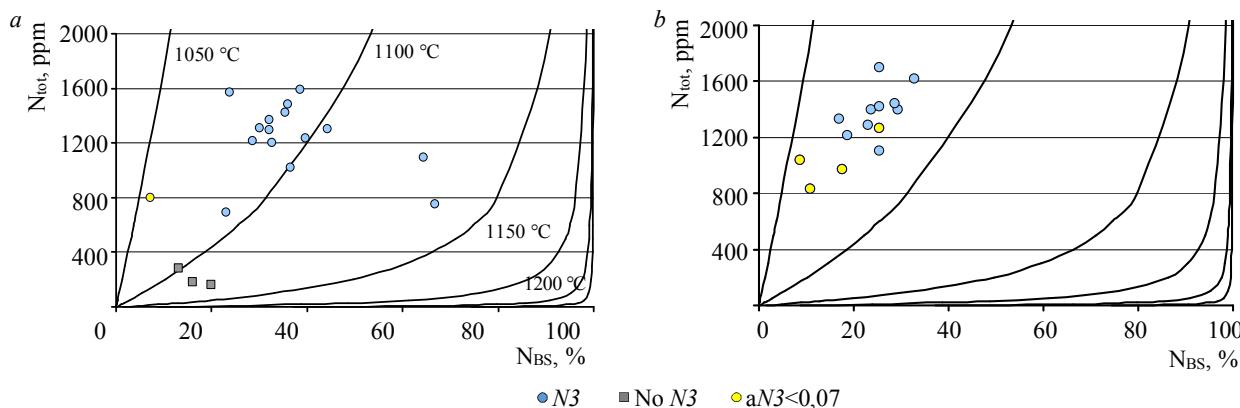


Fig.4. Distribution of diamond crystals of the octahedral habit of the Karpinsky-I (*a*) and Arkhangelskaya (*b*) pipes on the Taylor diagram with the separation of crystals by the presence of the  $N3$  system in the absorption spectrum

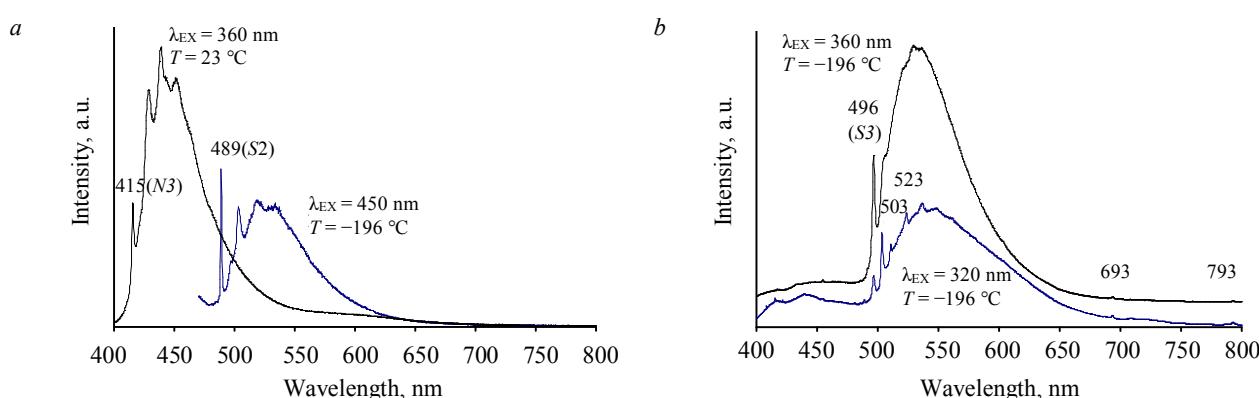


Fig.5. Luminescence spectra of crystals with the system  $N3$  and  $S2$ (489 nm) (*a*),  $S1$  and  $S3$  (*b*) under 77 K and excitation wavelength  $\lambda_{EX}$  of 320, 360, 450 nm

Table 2

**Occurrence frequency of luminescence systems in the spectra of diamond crystals of the Karpinsky-I and Arkhangelsk pipes**

Pipe	Luminescence system								
	<i>N3</i>	<i>S3</i>	<i>S2</i>	<i>*H3</i>	<i>*693nm</i>	<i>*700/787 nm</i>	<i>*926nm</i>	<i>*933nm</i>	<i>*H2</i>
	Recorded/Dominates**, %								
Karpinsky-I									
<i>N3</i> in absorption	98/67	74/31	64	12	23	58	41	14	2
No <i>N3</i> in absorption	39/5	94/92	45	30	80	93	85	1	3
Arkhangelskaya									
<i>N3</i> in absorption	98/83	51/17	89	15	27	62	35	25	2
No <i>N3</i> in absorption	54/27	76/73	68	34	90	95	62	6	5

\*At 77 K. \*\* Under excitation at 360 nm.

Thus, the presence of the *N3* system in the absorption spectrum is statistically related to other parameters: crystal morphology, thermal history, and nitrogen–hydrogen defect concentration. In addition, *N3* defects are active in luminescence and can be detected in crystals without a 415 nm peak in the absorption PL spectra. In the IR absorption spectrum of such crystals, there may be no *B* and *B'* absorption bands, but the *N3* system is registered in the PL. This sign is an indicator of the beginning of transformation of *A-B* stage defects. In a diamond with a high hydrogen concentration, most of the *N3* defects are transformed into *N<sub>3</sub>V* defects at the beginning of nitrogen transformation at the *A-B* stage. As a result, in cuboids and crystals of mixed habit containing a lot of hydrogen in micro inclusions, *N3* appears at a high degree of nitrogen aggregation. This effect explains the dominance of the blue luminescence of octahedra and the dissolution forms derived from them. It follows that dodecahedroids with *N3* blue luminescence are dissolution-exposed individual crystals of octahedral habit.

The study confirms the connection between the detected characteristics of the diamond crystal and the dominant mechanism of its growth. All crystals grown predominantly by the tangential mechanism (octahedra and octahedroids) have an appreciable fraction of nitrogen in the form of *B* defects and are distinguished by the presence of the *N3* system in the absorption and luminescence spectra. Crystals grown by the normal mechanism have a low degree of nitrogen aggregation and do not contain *N3* defects. Rounded dodecahedroids as the limiting form of dissolution [36] can be derivatives of both octahedral crystals and cuboids or crystals of mixed habit. However, individual crystals of a mixed habit are most often transformed upon dissolution into tetrahedroids with relics of cuboid surfaces. The work [1] presents images of crystals weighing more than 50 carats from the Grib deposit. As can be seen in the illustration [1, Fig.3.15], only two flat-faced octahedra are yellow, the rest are colorless or slightly colored, being the crystals of transitional forms and with signs of strong dissolution. This observation also confirms for large crystals, the revealed pattern of the complex thermal history of octahedral crystals and the high concentration of *N3* defects in them. Thus, even without a detailed study of crystal anatomy, it is possible to identify groups (populations) of crystals isolated by a set of features, which were formed under different conditions and differ in their post-growth history with respect to both natural annealing and the degree of dissolution.

**Conclusions.** Diamond crystals from the Arkhangelskaya and Karpinsky-I pipes have a unimodal distribution of nitrogen concentration of about 800-1100 and 900-1400 ppm, respectively, with a nitrogen aggregation degree of less than 20 %. The *N3* system of nitrogen-vacancy defects is detected in the absorption spectra of 30 % of crystals from the Karpinsky-I pipe and 52 % of the Arkhangelskaya pipe crystals. The *B'* band is in the spectra of 62 % and 78 % of samples from Karpinsky-I and Arkhangelskaya pipes, respectively. All octahedral habit crystals in these pipes have the *N3* system in their absorption spectra.



Crystals of octahedral habit formed earlier than dodecahedroids and cuboids, but post-growth processes changed their morphology less. The Karpinsky-I pipe is unique in its high proportion of crystals with an extremely low natural annealing temperature. Determining the range of thermal stability of defects in such diamonds is a task for further research. Differences in the thermal history and degree of dissolution of crystals different in morphology indicate a multistage formation of diamond from deposits of the north of the East European Platform.

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